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Extraction of aromatic sulfonates from aqueous solution with solid acid-immobilized magnetic graphene oxide. A solid acid-immobilized magnetic graphene oxide (SI-MAGGO) was prepared by a simple method and used to extract from aqueous solution four sulfonates with different ring structures: benzenesulfonic acid (SA), 2-nitrobenzenesulfonic acid (NBA), 4-aminobenzenesulfonic acid (ABA) and 4-methylbenzenesulfonic acid (MA). Kinetics studies indicated that the extraction of SA and ABA was relatively fast, and the maximum adsorption capacities reached for SA and ABA were 45.6 and 30.1 mg g⁻¹, respectively, within 10 min. In the meantime, the adsorption of SA and ABA from aqueous solution at equilibrium was evaluated by the Langmuir model and Freundlich model. The isotherm data of SA and ABA were fitted well by both models. The adsorption data fitted well for MA and NBA, with a tendency towards an inverse correlation between the adsorption capacity and molecular weight of the extractable sulfonate. These results suggested that the adsorption of SA and ABA is subject to "hydrophobic interaction", while MA and NBA are subject to "hydrophilic interaction".

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